

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

2-(1*H*-Indol-3-yl)-2'-(1*H*-indol-3-yl-methylene)acetohydrazide

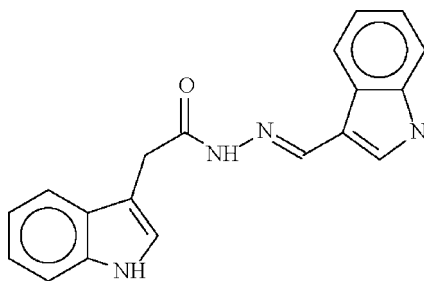
Hapipah M. Ali, Jamaludin Nazzatush Shimar and Seik Weng Ng\*

Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia  
Correspondence e-mail: seikweng@um.edu.my

Received 3 July 2007; accepted 4 July 2007

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  
R factor = 0.034; wR factor = 0.100; data-to-parameter ratio = 8.3.The molecules of the title compound,  $\text{C}_{19}\text{H}_{16}\text{N}_4\text{O}$ , are linked into a three-dimensional architecture by  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds.

## Related literature

For a related crystal structure, see Ali *et al.* (2007).

## Experimental

## Crystal data

 $\text{C}_{19}\text{H}_{16}\text{N}_4\text{O}$  $M_r = 316.36$ Monoclinic,  $Cc$  $a = 14.0350$  (2) Å $b = 14.4517$  (2) Å $c = 8.1372$  (1) Å $\beta = 102.193$  (1)° $V = 1613.23$  (4) Å<sup>3</sup> $Z = 4$ Mo  $K\alpha$  radiation $\mu = 0.08$  mm<sup>-1</sup> $T = 293$  (2) K

0.60 × 0.15 × 0.10 mm

## Data collection

Bruker APEXII CCD area-detector diffractometer  
Absorption correction: none  
12370 measured reflections2340 independent reflections  
2260 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.023$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.034$  $wR(F^2) = 0.100$  $S = 1.17$ 

2340 reflections

281 parameters

17 restraints

H atoms treated by a mixture of independent and constrained refinement

 $\Delta\rho_{\text{max}} = 0.38$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.26$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1N}\cdots\text{O1}^{\text{i}}$	0.88 (1)	1.98 (1)	2.848 (2)	174 (3)
$\text{N3}-\text{H3N}\cdots\text{O1}^{\text{ii}}$	0.87 (1)	1.97 (1)	2.808 (2)	163 (2)

Symmetry codes: (i)  $x + \frac{1}{2}, y - \frac{1}{2}, z$ ; (ii)  $x, -y + 1, z - \frac{1}{2}$ .Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2007).

The authors thank the University of Canterbury, New Zealand, for the diffraction measurements, and the University of Malaya (grant No. F0263/2007B) for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2473).

## References

- Ali, H. M., Nazzatush Shimar, J., Wan Jeffrey, B. & Ng, S. W. (2007). *Acta Cryst.* **E63**, o1807–o1808.
- Barbour, L. J. (2001). *J. Supramol. Chem.*, **1**, 189–191.
- Bruker (2005). *APEX2* (Version 2.0-2) and *SAINT* (Version 7.12a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Westrip, S. P. (2007). *publCIF*. In preparation.

**supplementary materials**

*Acta Cryst.* (2007). E63, o3459 [ doi:10.1107/S1600536807032655 ]

## 2-(1*H*-Indol-3-yl)-2'-(1*H*-indol-3-ylmethylene)acetohydrazide

H. M. Ali, J. Nazzatush Shimar and S. W. Ng

### Comment

A previous study reported 2'-[(5-chloro-1*H*-indol-3-yl)methylene]-2-(1*H*-indol-3-yl)acetohydrazide, a compound that is N–H···O hydrogen bonded into a three-dimensional network (Ali *et al.*, 2007). Although there are three possible donor sites, the site belonging to the indolyl ring that is linked to the methylene unit is not involved in the H bond interactions. A similar type of hydrogen bonding (Table 1) is found in the unsubstituted title compound, (I), (Fig. 1).

### Experimental

Indol-3-yl-acetylhydrazine (0.30 g, 1.6 mmol) and indol-3-ylcarboxaldehyde (0.20 g, 1.6 mmol) were heated in ethanol (50 ml) for 2 h. The solution was filtered; slow evaporation of the solvent yielded faint yellow needles of (I).

### Refinement

Anomalous dispersion was negligible and Friedel pairs were merged before refinement. All hydrogen atoms were located in difference Fourier maps, and were refined with distance restraints C—H = 1.00±0.01 Å and N—H = 0.88±0.01 Å. Their  $U_{\text{iso}}$  values were freely refined. Friedel pairs were merged.

### Figures

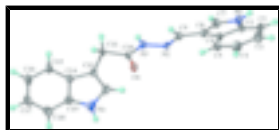


Fig. 1. View of (I) with displacement ellipsoids drawn at the 70% probability level, and H atoms as spheres of arbitrary radius.

## 2-(1*H*-Indol-3-yl)-2'-(1*H*-indol-3-ylmethylene)acetohydrazide

### Crystal data

C<sub>19</sub>H<sub>16</sub>N<sub>4</sub>O

$M_r = 316.36$

Monoclinic, *Cc*

Hall symbol: C -2yc

$a = 14.0350$  (2) Å

$b = 14.4517$  (2) Å

$c = 8.1372$  (1) Å

$\beta = 102.193$  (1)°

$V = 1613.23$  (4) Å<sup>3</sup>

$F_{000} = 664$

$D_x = 1.303$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 8692 reflections

$\theta = 2.8$ – $30.2$ °

$\mu = 0.08$  mm<sup>-1</sup>

$T = 293$  (2) K

Tapered needle, light yellow

$0.60 \times 0.15 \times 0.10$  mm

# supplementary materials

---

Z = 4

## Data collection

Bruker APEXII CCD area-detector diffractometer	2260 reflections with $I > 2\sigma(I)$
Radiation source: medium-focus sealed tube	$R_{\text{int}} = 0.023$
Monochromator: graphite	$\theta_{\text{max}} = 30.0^\circ$
$T = 173(2)$ K	$\theta_{\text{min}} = 2.1^\circ$
$\varphi$ and $\omega$ scans	$h = -19 \rightarrow 19$
Absorption correction: none	$k = -20 \rightarrow 20$
12370 measured reflections	$l = -11 \rightarrow 11$
2340 independent reflections	

## Refinement

Refinement on $F^2$	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.034$	$w = 1/[\sigma^2(F_o^2) + (0.0767P)^2 + 0.0497P]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.100$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.17$	$\Delta\rho_{\text{max}} = 0.38 \text{ e } \text{\AA}^{-3}$
2340 reflections	$\Delta\rho_{\text{min}} = -0.26 \text{ e } \text{\AA}^{-3}$
281 parameters	Extinction correction: none
17 restraints	Absolute structure: Friedel pairs were merged
Primary atom site location: structure-invariant direct methods	Flack parameter: ?
Secondary atom site location: difference Fourier map	Rogers parameter: ?

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.50000 (8)	0.58900 (8)	0.50001 (14)	0.0172 (2)
N1	0.86357 (11)	0.22337 (10)	0.56290 (18)	0.0207 (3)
N2	0.62106 (9)	0.44789 (8)	0.44976 (15)	0.0139 (2)
N3	0.55394 (9)	0.49082 (9)	0.32082 (15)	0.0143 (2)
N4	0.20100 (11)	0.47912 (12)	0.2027 (2)	0.0279 (3)
C1	0.87353 (11)	0.28098 (11)	0.7011 (2)	0.0176 (3)
C2	0.94407 (13)	0.28031 (13)	0.8506 (2)	0.0236 (3)
C3	0.93560 (13)	0.34678 (13)	0.9698 (2)	0.0260 (3)
C4	0.85926 (14)	0.41173 (13)	0.9415 (2)	0.0247 (3)
C5	0.79050 (12)	0.41325 (11)	0.7917 (2)	0.0193 (3)
C6	0.79757 (10)	0.34736 (10)	0.66815 (18)	0.0150 (3)
C7	0.78624 (12)	0.25238 (12)	0.44324 (19)	0.0204 (3)
C8	0.74261 (11)	0.32866 (10)	0.50057 (19)	0.0160 (3)
C9	0.66442 (11)	0.37875 (10)	0.39548 (18)	0.0154 (3)

C10	0.49922 (10)	0.56129 (9)	0.35533 (18)	0.0134 (2)
C11	0.43239 (10)	0.60328 (10)	0.20165 (19)	0.0152 (3)
C12	0.30093 (12)	0.48812 (12)	0.2490 (2)	0.0226 (3)
C13	0.33009 (11)	0.56780 (10)	0.18153 (18)	0.0154 (3)
C14	0.24281 (10)	0.61075 (10)	0.08736 (18)	0.0155 (3)
C15	0.22490 (11)	0.69156 (11)	-0.0099 (2)	0.0193 (3)
C16	0.12907 (13)	0.71310 (12)	-0.0859 (2)	0.0246 (3)
C17	0.05120 (13)	0.65631 (15)	-0.0636 (2)	0.0274 (4)
C18	0.06691 (12)	0.57637 (15)	0.0317 (2)	0.0276 (4)
C19	0.16355 (12)	0.55390 (12)	0.1053 (2)	0.0212 (3)
H1n	0.906 (2)	0.1812 (18)	0.551 (4)	0.046 (8)*
H3n	0.5469 (17)	0.4733 (16)	0.2169 (15)	0.017 (5)*
H4n	0.167 (2)	0.4364 (17)	0.242 (4)	0.041 (8)*
H2	0.9927 (18)	0.2299 (16)	0.860 (4)	0.041 (8)*
H3	0.9847 (16)	0.348 (2)	1.079 (2)	0.034 (7)*
H4	0.851 (2)	0.4539 (18)	1.034 (3)	0.041 (7)*
H5	0.7363 (14)	0.4593 (15)	0.772 (3)	0.027 (6)*
H7	0.7666 (19)	0.2195 (16)	0.334 (2)	0.026 (6)*
H9	0.6452 (19)	0.3595 (18)	0.2750 (15)	0.025 (6)*
H11A	0.4299 (18)	0.6713 (7)	0.218 (3)	0.022 (6)*
H11B	0.4619 (17)	0.5947 (18)	0.101 (2)	0.024 (6)*
H12	0.341 (2)	0.439 (2)	0.309 (4)	0.034 (7)*
H15	0.2789 (15)	0.7343 (17)	-0.021 (4)	0.032 (7)*
H16	0.117 (2)	0.7690 (13)	-0.159 (3)	0.037 (8)*
H17	-0.0162 (10)	0.6719 (19)	-0.124 (3)	0.032 (7)*
H18	0.0150 (16)	0.5337 (17)	0.050 (4)	0.037 (7)*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0198 (5)	0.0187 (5)	0.0121 (5)	0.0055 (4)	0.0011 (4)	-0.0006 (4)
N1	0.0239 (6)	0.0203 (6)	0.0181 (6)	0.0101 (5)	0.0049 (5)	0.0000 (5)
N2	0.0131 (5)	0.0155 (5)	0.0118 (6)	0.0026 (4)	0.0000 (4)	0.0025 (4)
N3	0.0158 (5)	0.0172 (5)	0.0088 (5)	0.0037 (4)	0.0001 (4)	-0.0002 (4)
N4	0.0196 (7)	0.0330 (8)	0.0278 (7)	-0.0075 (5)	-0.0029 (5)	0.0127 (6)
C1	0.0175 (6)	0.0186 (6)	0.0167 (7)	0.0046 (5)	0.0034 (5)	0.0022 (5)
C2	0.0198 (7)	0.0287 (8)	0.0207 (8)	0.0060 (6)	0.0005 (6)	0.0044 (6)
C3	0.0236 (7)	0.0316 (8)	0.0187 (8)	0.0000 (6)	-0.0046 (6)	-0.0001 (6)
C4	0.0268 (8)	0.0263 (8)	0.0187 (7)	-0.0013 (6)	-0.0007 (6)	-0.0057 (6)
C5	0.0210 (7)	0.0180 (7)	0.0181 (7)	0.0027 (5)	0.0022 (5)	-0.0031 (5)
C6	0.0147 (6)	0.0151 (6)	0.0149 (6)	0.0029 (4)	0.0024 (5)	0.0006 (5)
C7	0.0244 (7)	0.0199 (7)	0.0163 (7)	0.0065 (5)	0.0033 (5)	-0.0023 (5)
C8	0.0174 (6)	0.0157 (6)	0.0141 (6)	0.0040 (5)	0.0017 (5)	-0.0001 (5)
C9	0.0163 (6)	0.0170 (6)	0.0122 (6)	0.0026 (5)	0.0009 (5)	0.0001 (5)
C10	0.0123 (5)	0.0147 (5)	0.0122 (6)	0.0014 (4)	0.0004 (4)	0.0024 (5)
C11	0.0141 (6)	0.0176 (6)	0.0125 (6)	0.0027 (5)	-0.0005 (5)	0.0038 (5)
C12	0.0198 (7)	0.0254 (7)	0.0196 (7)	-0.0031 (6)	-0.0027 (6)	0.0074 (6)
C13	0.0150 (6)	0.0177 (6)	0.0123 (6)	0.0010 (5)	-0.0001 (5)	0.0013 (5)

## supplementary materials

---

C14	0.0149 (6)	0.0186 (6)	0.0121 (6)	0.0019 (5)	0.0010 (5)	-0.0014 (5)
C15	0.0189 (7)	0.0195 (6)	0.0180 (7)	0.0048 (5)	0.0005 (5)	0.0001 (5)
C16	0.0222 (8)	0.0244 (7)	0.0246 (8)	0.0086 (6)	-0.0010 (6)	0.0016 (6)
C17	0.0174 (7)	0.0389 (9)	0.0231 (8)	0.0073 (6)	-0.0018 (6)	-0.0001 (7)
C18	0.0157 (7)	0.0402 (9)	0.0250 (8)	-0.0026 (6)	-0.0001 (6)	0.0007 (7)
C19	0.0173 (6)	0.0279 (8)	0.0170 (7)	-0.0019 (5)	0.0005 (5)	0.0032 (6)

### *Geometric parameters (Å, °)*

O1—C10	1.2414 (19)	C7—C8	1.389 (2)
N1—C7	1.362 (2)	C7—H7	0.996 (10)
N1—C1	1.382 (2)	C8—C9	1.436 (2)
N1—H1n	0.876 (10)	C9—H9	1.000 (10)
N2—C9	1.2952 (18)	C10—C11	1.5220 (19)
N2—N3	1.3985 (16)	C11—C13	1.501 (2)
N3—C10	1.3405 (18)	C11—H11A	0.994 (10)
N3—H3n	0.868 (10)	C11—H11B	1.000 (10)
N4—C19	1.377 (2)	C12—C13	1.375 (2)
N4—C12	1.379 (2)	C12—H12	0.97 (3)
N4—H4n	0.883 (10)	C13—C14	1.4410 (19)
C1—C2	1.397 (2)	C14—C15	1.403 (2)
C1—C6	1.4171 (19)	C14—C19	1.415 (2)
C2—C3	1.388 (3)	C15—C16	1.392 (2)
C2—H2	0.991 (10)	C15—H15	0.996 (10)
C3—C4	1.406 (3)	C16—C17	1.409 (3)
C3—H3	1.002 (10)	C16—H16	0.998 (10)
C4—C5	1.386 (2)	C17—C18	1.383 (3)
C4—H4	0.993 (10)	C17—H17	0.994 (10)
C5—C6	1.404 (2)	C18—C19	1.400 (2)
C5—H5	0.998 (10)	C18—H18	0.991 (10)
C6—C8	1.443 (2)		
C7—N1—C1	108.94 (12)	N2—C9—H9	120.0 (16)
C7—N1—H1n	127 (2)	C8—C9—H9	117.2 (16)
C1—N1—H1n	124 (2)	O1—C10—N3	123.58 (13)
C9—N2—N3	112.15 (12)	O1—C10—C11	122.08 (13)
C10—N3—N2	120.32 (12)	N3—C10—C11	114.31 (13)
C10—N3—H3n	118.3 (16)	C13—C11—C10	111.28 (12)
N2—N3—H3n	121.4 (16)	C13—C11—H11A	107.0 (15)
C19—N4—C12	109.00 (14)	C10—C11—H11A	108.4 (14)
C19—N4—H4n	126 (2)	C13—C11—H11B	114.9 (15)
C12—N4—H4n	124 (2)	C10—C11—H11B	109.4 (15)
N1—C1—C2	129.47 (15)	H11A—C11—H11B	106 (2)
N1—C1—C6	108.12 (13)	C13—C12—N4	109.95 (14)
C2—C1—C6	122.40 (15)	C13—C12—H12	128.8 (18)
C3—C2—C1	117.02 (15)	N4—C12—H12	120.8 (18)
C3—C2—H2	128.1 (19)	C12—C13—C14	106.37 (13)
C1—C2—H2	114.8 (19)	C12—C13—C11	126.96 (13)
C2—C3—C4	121.48 (15)	C14—C13—C11	126.67 (13)
C2—C3—H3	118.9 (17)	C15—C14—C19	119.35 (14)

C4—C3—H3	119.6 (17)	C15—C14—C13	133.57 (14)
C5—C4—C3	121.32 (16)	C19—C14—C13	107.08 (13)
C5—C4—H4	119.3 (19)	C16—C15—C14	118.55 (15)
C3—C4—H4	119.2 (19)	C16—C15—H15	120.2 (17)
C4—C5—C6	118.54 (15)	C14—C15—H15	121.2 (17)
C4—C5—H5	121.0 (16)	C15—C16—C17	121.05 (16)
C6—C5—H5	120.4 (16)	C15—C16—H16	117.9 (17)
C5—C6—C1	119.21 (14)	C17—C16—H16	121.1 (17)
C5—C6—C8	134.37 (14)	C18—C17—C16	121.51 (15)
C1—C6—C8	106.42 (13)	C18—C17—H17	119.1 (17)
N1—C7—C8	110.21 (14)	C16—C17—H17	119.3 (17)
N1—C7—H7	121.4 (15)	C17—C18—C19	117.31 (16)
C8—C7—H7	128.4 (15)	C17—C18—H18	124.8 (18)
C7—C8—C9	122.39 (14)	C19—C18—H18	117.9 (18)
C7—C8—C6	106.28 (13)	N4—C19—C18	130.20 (16)
C9—C8—C6	131.06 (13)	N4—C19—C14	107.59 (14)
N2—C9—C8	122.73 (13)	C18—C19—C14	122.20 (16)
C9—N2—N3—C10	177.12 (13)	N2—N3—C10—C11	177.81 (12)
C7—N1—C1—C2	-177.43 (18)	O1—C10—C11—C13	-78.13 (17)
C7—N1—C1—C6	1.60 (18)	N3—C10—C11—C13	99.87 (15)
N1—C1—C2—C3	-179.51 (17)	C19—N4—C12—C13	-0.6 (2)
C6—C1—C2—C3	1.6 (3)	N4—C12—C13—C14	-0.1 (2)
C1—C2—C3—C4	0.2 (3)	N4—C12—C13—C11	179.88 (16)
C2—C3—C4—C5	-1.4 (3)	C10—C11—C13—C12	-20.7 (2)
C3—C4—C5—C6	0.8 (3)	C10—C11—C13—C14	159.29 (14)
C4—C5—C6—C1	0.9 (2)	C12—C13—C14—C15	-179.13 (17)
C4—C5—C6—C8	-178.63 (17)	C11—C13—C14—C15	0.9 (3)
N1—C1—C6—C5	178.76 (14)	C12—C13—C14—C19	0.75 (17)
C2—C1—C6—C5	-2.1 (2)	C11—C13—C14—C19	-179.23 (15)
N1—C1—C6—C8	-1.63 (17)	C19—C14—C15—C16	0.0 (2)
C2—C1—C6—C8	177.48 (15)	C13—C14—C15—C16	179.90 (17)
C1—N1—C7—C8	-0.93 (19)	C14—C15—C16—C17	1.2 (2)
N1—C7—C8—C9	174.55 (14)	C15—C16—C17—C18	-1.1 (3)
N1—C7—C8—C6	-0.11 (19)	C16—C17—C18—C19	-0.1 (3)
C5—C6—C8—C7	-179.41 (17)	C12—N4—C19—C18	-178.80 (19)
C1—C6—C8—C7	1.06 (17)	C12—N4—C19—C14	1.1 (2)
C5—C6—C8—C9	6.6 (3)	C17—C18—C19—N4	-178.77 (19)
C1—C6—C8—C9	-172.95 (16)	C17—C18—C19—C14	1.4 (3)
N3—N2—C9—C8	174.31 (13)	C15—C14—C19—N4	178.78 (15)
C7—C8—C9—N2	177.46 (15)	C13—C14—C19—N4	-1.13 (18)
C6—C8—C9—N2	-9.3 (3)	C15—C14—C19—C18	-1.3 (2)
N2—N3—C10—O1	-4.2 (2)	C13—C14—C19—C18	178.77 (16)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1N $\cdots$ O1 <sup>i</sup>	0.88 (1)	1.98 (1)	2.848 (2)	174 (3)
N3—H3N $\cdots$ O1 <sup>ii</sup>	0.87 (1)	1.97 (1)	2.808 (2)	163 (2)

# supplementary materials

---

Symmetry codes: (i)  $x+1/2, y-1/2, z$ ; (ii)  $x, -y+1, z-1/2$ .

Fig. 1

